

Solid State Characterization of Olmesartan medoximil Solid Dispersion and *in-silico* Formulation Design using Quality by Design Techniques Engendered by Definitive Screening Design

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ABSTRACT

Objectives: Olmesartan medoximil (OM) is employed for treating patients who are intolerant of ACE inhibitors. The challenge to the researchers is because of its poor oral bioavailability and poor solubility. The approach for this problem is to use a hydrophilic carrier in formulation of oro-dispersible tablet (ODT) which presents a suitable way to improve the bioavailability by using quality by design (QbD) techniques with design of experiments (DoE) using definitive screening design (DSD) which produce a robust and rugged formulation. **Methods:** The focus of the research was to formulate OM/PVP solid dispersion (SD) and formulation of an Oro dispersible tablet (ODT) by QbD techniques. The main focus of this research is to provide a rugged and robust formulation using QbD concept with the application of Definitive screening design for optimization. **Results:** The dissolution studies of OM/PVP K30 1:1% w/w showed full release within 30 min which may be attributed due to the hydrogen bond formation between OM and PVP K30 in the FTIR spectra which enhanced the solubility. The disintegration and dissolution results were found to be satisfactory and meeting the desired quality target product profile (QTPP). **Conclusion:** The

present research highlights a thorough understanding of the dosage form development with the knowledge of the critical risks involved in formulation to have an impact on critical quality attributes (CQAs). The critical material attributes (CMAs) were refined by DoE using definitive screening design (DSD) to develop design space.

Key words: Critical material attributes (CMAs), Critical quality attributes (CQAs), Definitive screening design (DSD), Quality by design (QbD), Design of experiments (DoE).

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INTRODUCTION

Olmesartan medoximil (OM) used an antagonist of angiotensin II receptor for treating high blood pressure. OM administered as a prodrug, has few drawbacks that it is completely de-esterified to Olmesartan as an active metabolite.¹ It acts by inhibiting the vasoconstrictor effects.² The usual recommended initial dose is 20 mg per day.³ Several preparations containing Olmesartan medoximil and other antihypertensives are available in the market.⁴ To overcome the drawbacks of low oral bioavailability the researcher aims to improve the dissolution of least water-soluble drugs by solubilization using some vehicle, reduction of particle size, solid dispersion and salt formation. The drugs polymorphism can be changed from crystalline to an amorphous state in solid dispersion system thus improving the solubility⁵ and also particle size reduction done for improved wettability.⁶ The drug solubility is also improved by the presence of the carrier which creates a microenvironment.⁷ Polyvinylpyrrolidone (PVP) is a synthetic high molecular weight polymer having linear groups of monomers of 1-vinyl-2-pyrrolidone exhibiting least toxicity with more hydrophilic property, physiological tolerance and enhances drug release and bioavailability⁸ as it is having universal solubility in hydrophilic and hydrooobic solvents.⁸

Taking all the above into account, the aim of this research work is to prepare solid dispersions of OM in PVP K30 using solvent evaporation technique and formulate it into an Oro Dispersible Tablet with QbD technology by using the design of experiments.

MATERIALS AND METHODS

Materials

Olmesartan medoximil, an active pharmaceutical ingredient, was obtained as gift sample from MSN Organics Pvt. Ltd., Hyderabad, India. PVP K30, Crosspovidone and starch purchased from Ascot Pharmachem Pvt. Ltd., Vadodara, India, Mannitol, Magnesium stearate and Aspartame purchased from Sigma Aldrich, Mumbai, India.

Methods

Preparation of Olmesartan medoximil solid dispersion

Solid dispersions with mass ratio of OM to PVP K30 ranging from 1:0.5 to 1:1.5 were formulated by solvent evaporation method.⁹ In brief, PVP K30 was dissolved in ethanol, followed by addition of OM. Ultrasonication at room temperature was done for about an hour to dissolve the drug completely¹⁰ and the remaining solvent was subjected to reduced pressure for evaporation. The resulting product was dried for 24 h at room temperature over anhydrous CaCl₂ desiccators in vacuum. The dried product was then pulverized and subjected through BSS 60# and stored in a desiccators^{8,11} and further evaluation done using Fourier transform infrared spectroscopy (FTIR Systronics, Ahmedabad, India).

Formulation of tablets with SD technology

Listed ingredients were weighed in required quantity and passed through suitable mesh. Binder solution was prepared by dissolving in starch in

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required amount of hot water and it is added to the drug-ingredient mixture to get uniform mass and passed through the suitable sieve and dried for 4 hrs in hot air oven at 45°C. Lubricants were added to the dried granules and micrometric properties were analysed and compressed with 5.2 mm concave punch. Formulated product was evaluated for physicochemical parameters.

Optimization of ODT parameters as per enhanced QBD

Optimization of the formulation using design of experiments with QbD approach,¹² QRM¹³ and knowledge management¹⁴ gives a robust formulation throughout the life cycle of the product.

Identification of CMA and CQA with justification

CQAs are linked with the drug substance, drug product and excipients.¹⁵ Relevant CQAs were identified based on the experience and prioritized through QRM and experimentation was done to assess the extent of variation of CMA's impact on the CQA.

Risk assessment of material attributes by QRM is a scientific approach that help in identifying CMA and CPP which show effect on product CQAs along with justification (Table 1). Experimentation was done using Design of Experiments (DoE) software to refine the list further to

evaluate the importance of individual variables and interactions to gain a higher degree of understanding.¹⁶

Optimization of material attributes and development of design space

Based on Initial Risk Matrix Analysis (IRMA), formulation understanding experiments viz DoE were implemented for the formulation. The effect of every independent CMAs on dependent CQAs (e.g. disintegration and dissolution) were analysed for establishment of Design space (DS) through timely evaluation of CQA which were modelled out with the target of achieving quality product. Definitive screening design (DSD) was used (Table 2) for optimization procedure for establishment of DS, because the design offers three levels and provides main effects estimation which were unbiased through any second order effects and requires only one trial greater than twice as many trials as there are factors and eliminates confounding effects on any pair of second order models thus favouring optimization of material attributes to achieve desired CQA. DSD was done using Design-Expert® software (Version 12, Stat-Ease Inc., Minneapolis, MN).¹⁷ Depending on IRMA, DoE were implemented for formulaion having higher risk priorities.¹⁸

RESULTS

The present work was aimed to prepare ODT for Olmesartan medoximil solid dispersion to increase bioavailability. Among the various approaches involved, ODT approach was selected as they are easy to fabricate and thereby enhancing the absorption of the drug. For this purpose, wet granulation technique was used with different excipients.

Drug-carrier interaction studies

FT-IR spectra helps in interpreting the interaction between the drug and other materials as shown in (Figure 1A and 1B). The spectra of pure OM showed bands at 3398.91 cm⁻¹ owing to N-H stretch, at 1706.98 cm⁻¹ owing to C=O stretching, The spectra also showed bands at 1225.39 cm⁻¹ owing to C-N bending. In the region, the 3398.91 cm⁻¹ NH stretching vibration peak of OM disappeared in the SD. It seems that there is a formation of intermolecular hydrogen bond between -NH of OM and -C=O of PVP. Thus, the appearance of characteristic absorption bands of OM and the solid dispersion containing OM showed no interaction between the OM and excipients.

Formulation Development Study

A Definitive Screening Design (DSD) with four factors and three levels were chosen to optimize varied response variables. The factors studied were decoded to allow for the ANOVA study. The relationship between factors and responses were elucidated by using counter plots. The Figure 2A indicates the ANOVA for Response Surface Quadratic for disintegration

Table 1: Initial Risk Assessment of the Material Attributes with Justification.

CMA	CQA	Target	Justification
Carrier	1. Dissolution	NLT 90% at 30 min	1. Carrier amount has direct impact on formation of solid dispersion thus inturn effects dissolution. Thus, the risk is medium.
	2. Disintegration	Less than 30 sec	2. Carrier amount has negligible direct affect on tablet disintegration time. Therefore, risk is low
Binder	1. Dissolution	NLT 90% at 30 min	1. Binder concentration has direct affect on dissolution. Therefore risk is high.
	2. Disintegration	Less than 30 sec	2. Binder concentration can affect the time of disintegration of tablet. Therefore risk is high
Disintegrant	1. Dissolution	NLT 90% at 30 min	1. Excess/Low level of disintegrant may affect the dissolution of tablet. Thus, the risk is high
	2. Disintegration	Less than 30 sec	2. Excess/Low level of disintegrant level may affect the tablet to disintegrate fastly/slowly. Therefore risk is high
Magnesium Stearate	1. Dissolution	NLT 90% at 30 min	1. Higher lubrication because of higher lubricant may slow dissolution. Therefore, risk is medium.
	2. Disintegration	Less than 30 sec	2. Magnesium stearate exhibits best lubrication properties, however it has less likely impact on disintegration time. Therefore risk is low.

Table 2: DSD study to investigate CMA and CQA.

Factors: CMA		Levels		
		-1	0	+1
A	Carrier	10	20	30
B	Disintegrant	10	20	30
C	Binder	1	2	3
D	Lubricant	1	3	5
Responses	Goal	Ranges Acceptable		
Y1	Disintegration time	Range 20-30		
Y2	Dissolution 15 min	Range 60-80		
Y3	Dissolution 30 min	Range 80-100		

was found to be significant ($P<0.05$) indicating fitness of the chosen equation $\text{Sqrt}(\text{Disintegration})=+4.89+0.6466A-1.46B+2.42C+1.54A^2$ and Contour plot for disintegration indicates an interaction between the critical factors and response variables respectively. The 2D graph for disintegration indicates the effect of carrier and disintegrant on disintegration time. At high levels of disintegrant, the disintegration time decreases and carrier does not have much influence on disintegration time.

The Figure 2B indicates the ANOVA for Response Surface Quadratic for % drug release profile at 15 mins (Q15) was found to be significant ($P<0.05$) indicating fitness of the chosen equation $\text{dissolution at 15 min}=+66.59+4.80A-1.80C-5.88A^2-5.88C^2$ and Contour plot for % drug release profile at 15 mins (Q15) indicates an interaction between the critical factors and response variables respectively. The 2D graph for Q15 indicates that effect carrier and disintegrant on %drug release at 15minutes. As the carrier level increases from 10mg to 20mg dissolution increases and remains constant when the carrier level further increases.

The Figure 2C indicates the ANOVA for Response Surface Quadratic for % drug release profile at 30 mins (Q30) was found to be significant ($P<0.05$) indicating fitness of the chosen equation $\text{dissolution at 30 min}=+95.33+6.20A+1.90 B-2.70C-3.13AB-10.13 A^2$ and Contour plot

for % drug release profile at 30 mins (Q30) indicates an interaction between the critical factors and response variables respectively. The 2D graph for Q30 indicates that effect carrier and disintegrant on %drug release at 30 minutes. As the carrier level increases from 10mg to 20mg dissolution increases and remains constant when the carrier level further increases. The dissolution % slowly increases as the amount of disintegrant increases from 15mg to 20mg. Disintegration and Dissolution (Q15, Q30, % drug release at 15 and 30 minutes) were taken as response variables (CQAs) and the corresponding results were shown (Table 3).

The experimental results for disintegration and dissolution at 15 min and 30 min when compressed at hardness 8 kpa and thickness 2.5 mm

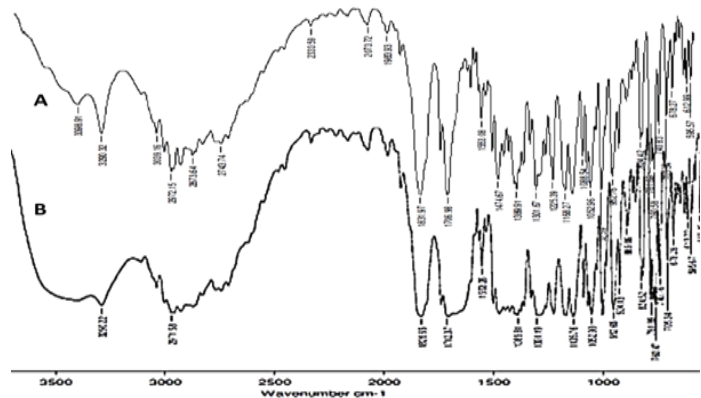


Figure 1: A. Fourier transform infrared spectra of Olmesartan medoximil B. Spectra of FTIR showing peaks for solid dispersion of OM with PVP K30.

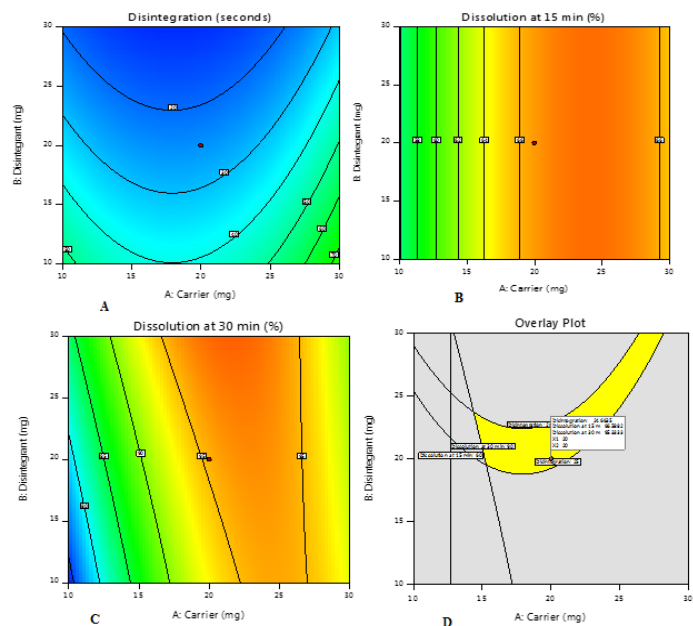


Figure 2: A. Effect on disintegration time due to carrier and disintegrant B. Effect on dissolution at 15 min due to carrier and disintegrant C. Effect on dissolution at 30 min due to carrier and disintegrant D. Overlay plot showing the Design Space.

Table 3: Experimental results to study carrier and excipients.

Run	Factors				Responses		
	A:Carrier	B:Disintegrant	C:Binder	D:Lubricant	Disintegration	Dissolution at 15 min	Dissolution at 30 min
	mg	mg	mg	mg	Seconds	%	%
1	30	30	3	1	68	58	86
2	30	30	1	3	8	64	94
3	30	10	3	5	127	58	89
4	10	20	3	1	50	50	78
5	10	30	1	1	7	56	88
6	20	20	2	3	21	70	100
7	10	30	2	5	12	55	82
8	20	30	3	5	48	58	92
9	10	10	3	3	97	47	73
10	20	10	1	1	10	60	94
11	30	20	1	5	21	61	95
12	10	10	1	5	35	48	74
13	30	10	2	1	72	63	93

Table 4: Final optimized formulation.

Ingredients (mg)	mg/tablet
Intragranular Portion	
Pure drug	20
PVP K30 (carrier)	20
Crosspovidone (disintegrant)	20
Mannitol (diluent)	34
Starch Rx 1500 (binder as paste)	2
Aspartame (sweetener)	1
Extragranular Portion	
Magnesium stearate (lubricant)	3
Total weight	100

indicate that all weight variation, thickness, friability and hardness were within the permissible limits of USP.

Thus, a composition having desired level of excipients is mentioned (Table 4). The optimized formulation has bulk density 0.295g/ml, angle of Repose 25°.56, Tapped Density 0.335 g/ml, Hausner's ratio 1.15 and Carr's index 11.94 %. The optimized formulation blend is compressed in to the tablets and quality control tests were done. The disintegration and dissolution results were found to be satisfactory and meeting the desired quality target product profile (QTPP).

DISCUSSION

The drug OM shows poor oral bioavailability due to its poor solubility which leads to reduced therapeutics efficacy. The thought-provoking aspect here is to enhance the bioavailability through the use of solid dispersion technique where a hydrophilic matrix polymer in the form of solvent was mixed with hydrophobic rug making it to form a dispersion.¹⁹ Ethylalcohol is used as a solvent to solubilize the drug²⁰ and it can be removed easily through evaporation. Ultrasonication was performed to simulate the homogenization effect and to dissolve the drug completely¹⁰ because of the reduction of the particle size and increase in surface area. In the formed solid dispersion, the drug can be present as molecularly dispersed,^{19,20} amorphous¹⁹ or microcrystal state²¹ with high energy. Once the solid dispersion was prepared, it was analysed to assess the physical state of the drug. It was found that the drug was molecularly dispersed in the PVP when it was used in the ratio of 1:1 when compared with the literature results in which 1:6 ratio of drug and PEG-4000 was used.²² The amount of molecularly dispersed drug depends on the ratio of the drug and the polymer.²³ The solid dispersion formed was characterized using FTIR which shows no interaction between drug and polymer and formation of intermolecular hydrogen bond between -NH of OM and -C=O of PVP²⁴ during the solvent evaporation process confirms the molecular dispersion of drug into the polymer matrix and further the formed SD used as an active ingredient for the formulation of tablets.²⁵ These findings are important, in the light of QbD concept in the pharmaceutical formulation development since QbD requires knowledge of the influence of formulation components (material attributes) on the final product properties and identification of interactions in order to select the optimal formulation. Definitive screening design was chosen to optimize the formulation with four factors viz carrier (X1), disintegrant (X2), binder (X3) and lubricant (X4) and three responses viz disintegration (Y1), dissolution at 15 min (Y2) and dissolution (Y3). Dissolution medium used contains pH 6.8 phosphate buffer in type 2 apparatus operated at a speed of 50 rpm for 30 min and analysed at 257 nm as per USP. One of the main goals in SD

processing is to achieve an improvement in the dissolution rate of OM and the results obtained are comparatively equivalent with the literature reports in terms of drug release of OM with polaxmer 407,⁹ faster than OM with cyclodextrin.²⁶ Faster than OM with VA64²⁷ and faster than OM with soluplus.²⁷ Dissolution studies revealed that the formulations are in accordance to the USP requirements for OM immediate release tablets. The 2D counter graph indicates that inclusion of PVP K30 in the mixture influences positively the amount of OM released (%) after 15 min and 30 min, but only up to a certain amount, which corresponds to approximately 1:1 ratio of OM to PVP K30 and were confirmed by the release from formulation labelled as run 6 (containing approximately 1:1 ratio of both OM and PVP K30), which had the fastest onset.

On the basis of the obtained results, it is clear that immediate release solid dispersions of OM can be successfully prepared using the solvent evaporation technique. Regardless of their composition, all critical material attributes display an improvement in the release rate and the percentage of OM released, compared to the pure OM which showed 25% release in 30 min. The study demonstrated that interactions between the mixture components (SD and excipients), or the quadratic effects of the components, play a significant role in overall influence on the OM release rate. After analysing the data obtained through ANOVA using the final equation, the desired goal for each CMA and CQA were chosen in Design Expert software. The goal selection starts at initial point and goes to a maximum where it was required to select a region where requirements meet the CQA. Graphical optimization showed the area of feasible response values in yellow colour. The gray shaded region indicates that the optimization criteria did not meet as represented in Figure 2D. The yellow region constituted a possible design space for robust and rugged Formulation. From the results obtained through design space, the risk assessment of the material attributes was updated as low for all the CQA. Thus the study design successfully predicted values of independent variables for optimization of the dosage form. All the experimental values were in close agreement with the predicted values indicating goodness of fit for the model.²¹ Application of DSD has enabled the quantitative description of the relations between mixture components. These findings can serve as a foundation for the optimization of the future OM immediate release products.

CONCLUSION

The possible enhancement of dissolution rate was due to solid dispersion containing 1:1 mass ratio of OM:PVP K30. The appearance of hydrogen bonding between the >NH of OM and -C=O of PVP K30 in SD characterised by FTIR leads to decrease in crystallinity and the primary reason for the marked increase of dissolution rate. The results showed that OM-PVP K30 SD formulated using solvent evaporation served as a means of increasing OM dissolution rates. The results of QRM study of ODT formulation proved that statistical tools of QbD helped in achieving the quality product throughout the lifecycle.

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CONFLICT OF INTEREST

The authors declare no conflict of interest

ABBREVIATIONS

CMA: Critical material attributes; **CQA:** Critical Quality Attributes; **QTPP:** Quality Target Product Profile; **ODT:** Oro dispersible tablet; **DSD:** Definitive Screening Design; **FTIR:** Fourier Transform Infrared

Spectroscopy; **QbD**: Quality by Design; **PVP**: Polyvinylpyrrolidone; **DoE**: Design of Experiments; **QRM**: Quality Risk Management; **SD**: Solid Dispersion; **ANOVA**: Analysis of Variance.

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