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A Simple HPLC-MS/MS Method for Quantification of Naltrexone and 6-beta Naltrexol: An Application to Effect of Uremic Toxins on Metabolic Reduction

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ABSTRACT

Objective: This study aimed to 1) develop and validate a simple, accurate and reproducible high-performance liquid chromatography-tandem mass spectrometry method for determination and quantification of naltrexone and its reduced metabolite (6-beta naltrexol); 2) assess the effect of uremic toxins on metabolic reduction. Methods: Sample preparation was conducted through liquid-liquid extraction of analytes spiked in 10 mM Tris-HCl incubation buffer (pH 7.4) using methyl tert-butyl ether. The chromatographic separation of analytes was achieved using Inert Sustain C_{18} (2.1 x 50 mm, 5 μm) analytical column under isocratic elution of solvent A (water containing 0.1% formic acid) and solvent B (methanol containing 0.1% formic acid) over a run time of 5 min. The analytes were detected by multiple reaction monitoring with electrospray ionization in the positive mode. Results: Excellent linearity was observed for all analytes over the concentration ranges of 100-8,000 ng/mL for naltrexone and 10-800 ng/mL for 6-beta naltrexol. The intra- and inter-day accuracy and precision for analytes were within ±15.0%. The method is accurate and precise and novel in its application

to *in vitro* experiments assessing the effect of uremic toxins on metabolic reduction of naltrexone, which showed that indoxyl sulfate (100 μ M) is a significant inhibitor of naltrexone reduction. **Conclusion:** The developed method is reproducible and accurate and well suited for conducting enzyme kinetic studies assessing metabolic reduction.

Key words: Naltrexone, 6-beta naltrexol, Reduction, LC-MS, Enzyme kinetics, Uremic toxins.

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INTRODUCTION

Naltrexone is an opioid antagonist used in the treatment of opiate dependence and alcoholism.¹ It works by binding and inhibiting of beta opioid receptors located in brain. Naltrexone is a highly metabolized drug that undergoes hepatic reduction at the carbonyl moiety to produce primarily 6-beta naltrexol metabolite.^{2,3} Naltrexone reduction is principally mediated by hepatic Aldo-keto Reductase (AKR) isoforms located in cytosol.³⁻⁵ While the 6-alpha metabolite has not been recognized in hu

mans, the beta isoform is pharmacologically active and inhibits opioid receptors to levels lower than that of the parent drug. Additionally, the plasma half-life of 6-beta naltrexol is longer than that of naltrexone and it is detected in plasma at higher levels, which may contribute to the clinical effects observed with naltrexone treatment.⁶

As naltrexone is chiefly metabolized by reduction, the drug may act as a candidate of reductase enzyme probe substrates. Therefore, naltrexone reduction may be utilized to study the effect of disease state and therapeutic treatments on metabolic reduction to gain a better understanding of this metabolic pathway. For instance, kidney disease has been recently shown to alter drug reduction *in vitro*.⁷ However, the mechanisms behind this alteration are unclear. In order to assess factors contributing to alter metabolic reduction, it is necessary to have a robust and validated analytical method suitable for the quantitative determination of 6-beta naltrexol metabolite generated from *in vitro* experiments.

Because of inherent selectivity and specificity of Mass Spectrometry (MS), multiple analytical methods have been developed utilizing High

Performance Liquid Chromatography (HPLC) coupled with MS detection for the determination of naltrexone, with only few assays have quantified naltrexone and 6-beta naltrexol simultaneously.⁸⁻¹³ Additionally, these methods were validated in biological fluids such as serum, plasma and blood of animals or human. Up to our knowledge, there were no reports specifically designed to quantify 6-beta naltrexol following *in vitro* enzyme kinetic studies.^{4,5} Thus, we developed and partially validated a simple and rapid HPLC-MS/MS assay for the simultaneous determination of naltrexone and 6-beta naltrexol metabolite following *in vitro* incubations. The method was used to examine, for the first time, the effect of uremic toxins on cytosolic reduction of naltrexone.

MATERIALS AND METHODS

Chemicals and Reagents

Naltrexone ($C_{20}H_{23}NO_4$) and 6-beta naltrexol ($C_{20}H_{25}NO_4$) were purchased from Toronto Research Chemicals (Canada) and Sigma (USA), respectively. Colchicine ($C_{22}H_{25}NO_6$, used as internal standard), hippuric acid, P-cresol, magnesium chloride, 3-carboxy-4-methyl-5-propyl-2-furanpropanoic acid (CMPF) and testosterone were purchased from Acros (USA). Tris (hydroxymethyl) aminomethane (Trizma base) was obtained from Bio Basic (Canada). Creatinine, indole-3-acetic acid and NADPH were purchased from Combi-Blocks Inc (USA). Indoxyl sulfate potassium salt was purchased from Santa Cruz Biotechnology (USA). Chemical solvents were purchased from Thermo Fisher Scientific (USA)

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or BBC Chemicals (USA). All chemicals were with the highest purity and analytical grade available. Nitrogen gas (ultra-pure, >99.9%) was produced by a Parker nitrogen generator (USA). Pooled human liver cytosol (20 mg/mL) was obtained from Gibco/ThermoFisher Scientific (USA).

Equipment and HPLC-MS/MS Conditions

Liquid chromatography was performed with an Agilent 1200 Series HPLC system (Agilent Technologies, CA, USA) including an autosampler and high-performance binary pump. Chromatographic separation of the samples was achieved with Inert Sustain C_{18} (2.1 × 50 mm, 5 µm, GL Sciences, Japan) analytical column. Analytes were separated under isocratic mobile phase over flow rate of 200 µL/min and composed of 20% solvent A (water containing 0.1% formic acid) and 80% solvent B (methanol containing 0.1% formic acid). The total run time was 5 min. The autosampler was maintained at 10°C and the column temperature was held at 40°C.

MS/MS detection was performed on API- 3200 triple quadruple mass spectrometer (Applied Biosystems/ MDS Sciex) equipped with a heated electrospray ionization source. Analytes were detected in positive ionization mode using multiple reaction monitoring with a dwell time of 150 ms. The spray voltage was set to 5500V and the ion source temperature was set to 450°C. The sheath gas and auxiliary gas were set to 45 and 50 (arbitrary units), respectively. Collision gas pressure was set at 5 mTorr. The ion transitions were *m*/*z* 342.2 \rightarrow 322.8 for naltrexone (collision energy = 27V), *m*/*z* 344.2 \rightarrow 324.8 for 6-beta naltrexol (collision energy = 25V) and *m*/*z* 400.2 \rightarrow 152.1 for the internal standard colchicine (collision energy = 115V). Data were processed with Analyst Software (version 1.6.2, Applied Biosystems/ SCIEX).

Preparation of Calibration Standards and Quality Controls

Naltrexone was dissolved in methanol and 6-beta naltrexol was dissolved in DMSO to prepare 1.0 and 5.0 mg/mL stock solutions, respectively. These solutions were spiked into 10 mM Tris-HCl incubation buffer (pH 7.4) to create working solutions that were used to prepare calibration standards at concentrations of 100, 150, 300, 600, 1500, 3000, 6000 and 8000 ng/mL for naltrexone and 10, 15, 30, 60, 150, 300, 600 and 800 ng/ mL for 6-beta naltrexol. Three quality control (QC) samples (LQC, MQC and HQC) containing both analytes were made by spiking analyte stock and working solutions into Tris-HCl buffer to prepare three concentration levels of 400, 2000 and 7000 ng/mL for naltrexone and 40, 200 and 700 ng/mL for 6-beta naltrexol metabolite. Stock solutions were stored at -40°C, whereas calibration standards and QC samples were stored at 4°C.

Sample Preparation

The internal standard colchicine was initially prepared in methanol at stock concentration of 1.0 mg/L that was further diluted in Tris-HCl buffer to prepare a working solution containing 100 μ g/mL. A 200 μ L aliquot of sample (calibration standards, QCs, blanks, application samples) was spiked with 20 μ L internal standard solution and vortexed. Liquid-liquid extraction was carried out by adding 2 mL Methyl Tert-butyl Ether (MTBE) to samples followed by vigorous vortex-mixing at room temperature. Samples were then centrifuged at 3000 rpm for 15 min and the upper organic layer was then removed using a pipette and evaporated to dryness under vacuum at 45°C using CentriVap Console Mobile System (Labconco, USA). The dried residue was then reconstituted with 100 μ L of mobile phase and a 20 μ L aliquot was injected onto the LC-MS/MS system.

Assay Validation

Calibration and Linearity

Calibration curves were constructed using eight concentrations of naltrexone and its reduced metabolite (6-beta naltrexol). Each calibration level (Including blank sample with or without internal standard) was run in duplicate for three days, except for the Lower Limit of Quantification (LLOQ), which was run in 3-5 replicates. For each curve, the peak area ratios of the analyte to the internal standard were calculated and plotted against the nominal analyte concentration. Calibration curves were generated by 1/x weighted linear regression model that provided the best fit to data.

Accuracy and Precision

The accuracy and precision were assessed by analysis of six replicates of QC samples run on three days, with a total of n=18 samples at each QC level. Intra-day accuracy and precision were determined by analysing six replicates run at one day, whereas inter-day accuracy and precision were estimated using all 18 QC samples. Accuracy was expressed as (% bias) which reflects the measured concentration relative to the nominal concentration. The precision was expressed as relative standard deviation (% RSD). For the accuracy and precision, the acceptance criteria should be within 15% for the QC samples and within 20% for the LLOQ.

Extraction Efficiency and Matrix Effect

The extraction efficiency was assessed by comparing the peak areas of analytes spiked in the buffer before extraction with that of neat solutions spiked in the mobile phase, which was defined as 100% recovery. Matrix effect was evaluated by comparing the response of the analytes spiked in the buffer after sample processing with that of neat solutions spiked directly in mobile phase, representing 100% (no matrix effect). All experiments were conducted in four replicates at the three QC levels.

Application of the Method

The current method was applied by evaluating the impact of uremic toxins on metabolic reduction using naltrexone as probe substrate. The formation of 6-beta naltrexol metabolite in vitro was quantified after naltrexone reduction. Human hepatic cytosolic protein (0.5 mg/mL) was incubated with naltrexone (50 µM) and 5 mM MgCl, in 10 mM Tris-HCl buffer (pH 7.4). Incubations were conducted in the presence of uremic toxins such as hippuric acid, p-cresol, 3-indoxyl sulfate, indole-3-acetic acid, CMPF and creatinine at previously reported concentrations of 10 and 100 µM.14,15 Substrate concentration was selected based on Km value determined on preliminary experiments. Control incubations were performed with vehicles (DMSO or water) at a final concentration of 1% v/v. Testosterone (100 µM) that was previously shown to inhibit naltrexone reduction and was used as a positive control.5 The reaction mixture was pre-incubated for 5 min in a shaking water bath at 37°C prior to the start of the reaction by addition of 1 mM NADPH. Reactions lasted for 30 min and terminated by the addition of ice-cold acetonitrile. Samples were then mixed briefly and placed on ice for 10 min. The mixture was then centrifuged at 10,000 rpm for 15 min to pellet cytosolic protein. The supernatant was collected and the analytes were quantified as described above. Incubations were conducted in triplicates.

The formation rate (pmoles/mg protein/min) of 6-betal naltrexol metabolite reflecting the reductase enzyme activity was calculated from the measured concentrations. In the inhibition experiments, the formation rate was presented as percentage of control (vehicle) incubations. Statistical differences in the formation of metabolite were compared in the presence and absence of inhibitors using student's *t* test assuming equal variances. The *P* value of <0.05 was considered statistically significant. Analysis was performed using (GraphPad Software Inc., San Diego, CA).

RESULTS

Optimization of LC-MS/MS Conditions, Chromatography and Extraction Method

MS parameters (tube lens and collision energy) were optimized automatically to achieve optimum detection of the analytes with best sensitivity. Individual standard solutions (10 μ g/mL) of the analytes and internal standard were directly infused into the instrument equipped with ESI source operating in positive ionization mode. Optimum mass transitions were selected for each analyte and were used later for their quantification.

Chromatographic conditions were optimized with the aim of achieving the shortest run time and best peak shape. Various mobile phases' composition and ratios, different formic acid concentrations and variable flow rates were employed. Best peak shapes and high sensitivity were obtained when 0.1% formic acid was added to the mobile phase. The enhanced sensitivity is attributed to the better protonation of the analytes in the positive ion mode.

The sample preparation step of liquid–liquid extraction was employed to isolate the analytes from the buffer and reduce potential matrix effect associated with it. It also provides a step for preconcentration the sample if lower limit of detection is required. Multiple organic solvents (MTBE, diethyl ether, methylene chloride) were evaluated and MTBE was selected as it showed the highest extraction recovery compared to other solvents.

Chromatographic Separation

Representative chromatograms of LLOQ and study sample depicting the internal standard are displayed in Figure 1. The retention times were approximately 0.66, 0.65 and 0.88 min for naltrexone, 6-beta naltrexol and colchicine, respectively.

Assay Validation

The assay was found to be linear over concentration ranges of 100-8,000 ng/mL for naltrexone and 10-800 ng/mL for 6-beta naltrexol, with a correlation coefficient (r^2) ranged of (0.990-0.998) for all curves. The LLOQ for each calibration curve demonstrated acceptable accuracy and precision with intra- and inter-day values of RSD and bias were within $\pm 7.3\%$ and $\pm 11\%$ respectively, for both analytes (Table 1). The signal-to-noise was greater than 10:1 for both analytes.

The intra-day and inter-day accuracy (% bias) and precision (% RSD) were determined at naltrexone QC concentrations of 400, 2000 and 7000 ng/mL and 6-beta naltrexol concentrations of 40, 200 and 700 ng/mL. Assay bias ranged from -14.7 to 15%, while the RSD ranged from 1.5 to 13.2% for all QC levels at intand inter-day assessments (Table 1). In all cases, bias and RSD values were within ±15% for all analytes.

The mean extraction efficiency of both analytes at all QC levels ranged from 73.2% to 114.8%. The matrix effect was minor as the measured concentrations of both analytes were deviated by 23.8% to 37.1% from neat samples at all QC levels (Table 2). The RSD values were within $\pm 15\%$ for all samples.

Assay Application

The current assay was successfully applied by evaluating the effect of multiple uremic toxins at 10 and 100 μ M on naltrexone reduction. The results revealed that indoxyl sulfate at 100 μ M inhibits significantly naltrexone reduction by 27% as compared to the vehicle control. Similarly, the formation rate of 6-beta naltrexol was decreased by 26% in the presence of the positive control testosterone (100 μ M), Figure 2. The other investigated uremic toxins including hippuric acid, p-cresol, indole-

3-acetic acid, CMPF and creatinine did not have an effect on naltrexone reduction.

DISCUSSION

In here, we aimed at developing and validating a simple and robust LC-MS/MS method for quantification of naltrexone and 6-beta naltrexol metabolite to support *in vitro* studies assessing phase I metabolic reduction. This assay was partially validated according to the U.S. Food and Drug Administration (FDA) guidelines for bioanalytical method validation.¹⁶ The method is simple, accurate and precise and was applied to inhibition enzyme kinetic studies of naltrexone reduction and found for the first time that indoxyl sulfate may inhibit metabolic reduction.

Carbonyl reduction is the primary Phase I metabolic pathway for xenobiotics bearing a carbonyl group. Naltrexone reduction to its primary alcohol metabolite (6-beta naltrexol) represents an example of such type of metabolism.³ A previous study characterized the reductase isoforms contributing to naltrexone reduction found that AKR1C1, AKR1C2 and AKR1C4 were able to metabolize naltrexone, with the highest efficacy was for AKR1C4.⁴ This suggests that the formation of 6-beta naltrexol may act as an indicative of reductase activity, which motivated the development of the current LC-MS/MS assay.

 Table 1: Intra-and inter-day accuracy (% bias) and precision (% RSD) for

 LLOQ and quality control samples of analytes.

Analyte	Level	Nominal conc. (ng/mL)	Intra-day ¹	Inter-day ²		
			% bias	% RSD	% bias	% RSD
Naltrexone	LLOQ	100	-3.8	5.5	5.4	-5.75
	LQC	400	15.0	1.5	14.1	4.5
	MQC	2000	3.5	7.0	4.6	13.1
	HQC	7000	0.17	5.8	0.3	7.1
6-beta naltrexol	LLOQ	10	-7.3	6.5	-11.0	9.75
	LQC	40	-14.7	9.2	-1.3	13.2
	MQC	200	3.9	7.6	1.7	11.8
	HQC	700	-4.6	4.6	-2.1	4.4

¹Five replicates for LLOQ; six replicates for LQC, MQC, HQC

²Eleven replicates for LLOQ; eighteen replicates for LQC, MQC, HQC

Table 2: Extraction efficiency and matrix effect of analytes in the incubation buffer.

Analyte	Level	Nominal concentration (ng/mL)	Extraction efficiency (%, mean± RSD)	Matrix effect (%, mean± RSD)
Naltrexone	LQC	400	101 ± 3.7	76.2 ± 2.7
	MQC	2000	111.9 ± 2.6	72.3 ± 2.7
	HQC	7000	114.8 ± 3.5	68.9 ± 2.9
6-beta naltrexol	LQC	400	74.4 ± 11.8	62.9 ± 15.0
	MQC	2000	80.3 ± 3.9	71.9 ± 5.4
	HQC	7000	73.2 ± 3.9	71.0 ± 1.5

RSD: relative standard deviation.



Figure. 1: Representative chromatograms of incubation buffer at LLOQ (left panel, A-C), and a cytosolic sample incubated with indoxyl sulfate (right panel, D-F). (A) naltrexone: 100 ng/mL; (C, F) internal standard colchicine; (B) 6-beta naltrexol: 10 ng/mL; (D) naltrexone: 7,920 ng/mL; (E) 6-beta naltrexol: 360 ng/mL.

While there are different analytical sensitive assays have been developed for quantification of naltrexone and its metabolites, the presence of an online sample cleaning step with column switching9 or the long run time¹⁰ may act as limitations. The current method, while it is less sensitive, it utilized a simple extraction procedure and has a short run time of 5 mins. Additionally, this assay was specifically developed to quantify 6-beta naltrexol after in vitro incubations. The previous reported methods11,13 employed a relatively large sample volume (1 mL) compared to our method which used 200 µL sample volume. This is important, especially in enzyme kinetic studies since incubation volumes are relatively small. The linear range selected for this study is relatively wide and applicable for enzyme kinetic investigations. For instance, the concentrations of the metabolite obtained from the in vitro studies were within the calibration curve. The results of accuracy and precision experiments demonstrate that the current developed method is satisfactory valid and reproducible. Furthermore, the process efficiency was consistent and reproducible for individual analyte at the three QC levels.

Several studies have shown that kidney disease alters selectively the functional expression of drug metabolizing enzymes.¹⁷ For example, the activity of hepatic reductase enzymes that mediate metabolic reduction is found to be decreased in the setting of impaired kidney function.⁷ However, the mechanisms behind this alteration are not well established. In here, we hypothesized that uremic toxins accumulated in the serum of kidney disease patients are implicated in this alteration. Therefore, the present method was applied by assessing the effect of various uremic toxins at 10 and 100 μ M on naltrexone reduction. The finding that indoxyl sulfate is an inhibitor of drug metabolism has also been reported earlier by Tsujimoto *et al.* who showed that metabolism of losartan in human liver microsomes mediated by CYP450 is inhibited by indoxyl sulfate in



Figure. 2: Effect of uremic toxins on the metabolic reduction of naltrexone in human liver cytosol. The y-axis represents the formation rate of the metabolite (6-beta naltrexol) which was calculated from the measured concentrations. ** indicates p value <0.01. n= 2 experiments.

a dose dependent manner.¹⁸ All other toxins tested in our study showed negligible effect on naltrexone metabolic reduction indicating that the effect on drug metabolism might be linked to specific uremic toxins. The latter observation is consistent in part to the results of previous studies that revealed that indole-3-acetic acid,¹⁴ creatinine,¹⁹ CMPF and hippuric acid¹⁵ did not have an effect on CYP450 activity. Overall, the results indicate that the current assay is suitable for enzyme kinetics studies pertain to metabolic reduction.

The study has limitations. The sensitivity of the method was lower than previous assays. However, the range of the metabolite concentration selected was relevant to the purpose of metabolite formation in the presence of inhibitors. Also, the study did not assess the kinetic parameters of inhibitors such as IC₅₀ or Ki, which might be explored in future investigations.

CONCLUSION

A simple, rapid, accurate and reproducible LC-MS/MS method for the determination of naltrexone and 6-beta naltrexol metabolite has been developed and validated. These characteristics along with the short run time and the selectivity associated with LC-MS/MS analysis renders this assay well suited for conducting enzyme kinetic studies assessing metabolic reduction (i.e., naltrexone).

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CONFLICT OF INTEREST

The authors declear no conflict of interest.

ABBREVIATIONS

AKR: Aldo-Keto Reductase; **CMPF:** 3-carboxy-4-methyl-5-propyl-2-furanpropanoic acid; **FDA:** Food and Drug Administration; **HPLC:** High Performance Liquid Chromatography; **HQC:** High Quality Control; **LLOQ:** Lower Limit of Quantification; **LQC:** Low Quality Control; **MS:** Mass Spectrometry; **MTBE:** Methyl Tert-Butyl Ether; **MQC:** Middle Quality Control; **QCs:** Quality Control samples; **RSD:** Relative Standard Deviation; **r2:** Correlation Coefficient.

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